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Supporting Information

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# **Zipped-up Chain-type Coordination Polymers: Unsymmetrical Amide-containing Ligands Inducing *b*-Sheet or Helical Structures**

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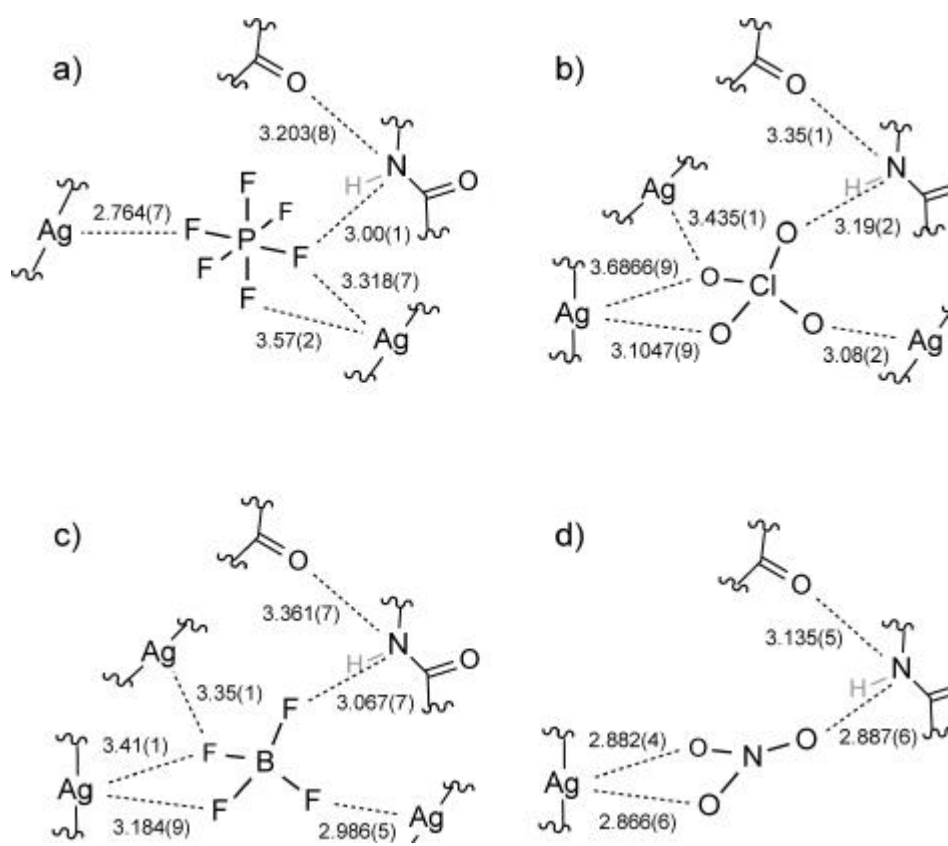


Figure S1. Schematic views of the anion environments in (a)  $\{[\text{Ag}(4\text{-pia})](\text{PF}_6)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{PF}_6^-$ ), (b)  $\{[\text{Ag}(4\text{-pia})](\text{ClO}_4)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{ClO}_4^-$ ), (c)  $\{[\text{Ag}(4\text{-pia})](\text{BF}_4)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{BF}_4^-$ ) and (d)

$\{[\text{Ag}(4\text{-pia})](\text{NO}_3)\}_n$  (**1**  $\dot{\text{E}}$   $\text{NO}_3^-$ ). Numbers exhibit the distances ( $\text{\AA}$ ) among atoms. Expressions for double bonds in  $\text{ClO}_4^-$  and  $\text{NO}_3^-$  anions are omitted.

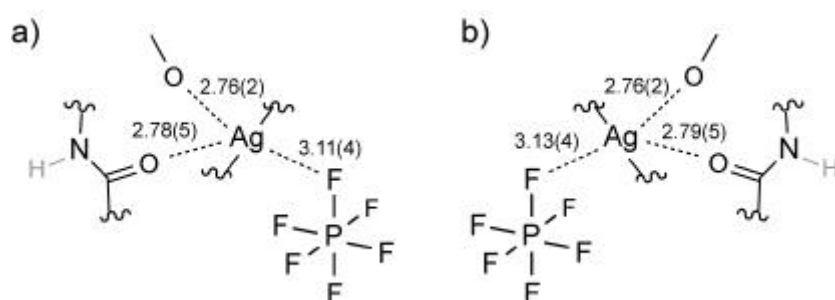


Figure S2. Schematic views of the  $\text{PF}_6^-$  environments in (a)  $\{[\text{Ag}(4\text{-pmna})](\text{PF}_6)\cdot\text{MeOH}\}_n$  (**4a**  $\dot{\text{E}}$   $\text{PF}_6^- \cdot \text{MeOH}$ ) and (b)  $\{[\text{Ag}(4\text{-pmna})](\text{PF}_6)\cdot\text{MeOH}\}_n$  (**4b**  $\dot{\text{E}}$   $\text{PF}_6^- \cdot \text{MeOH}$ ). Numbers exhibit the distances ( $\text{\AA}$ ) among atoms.

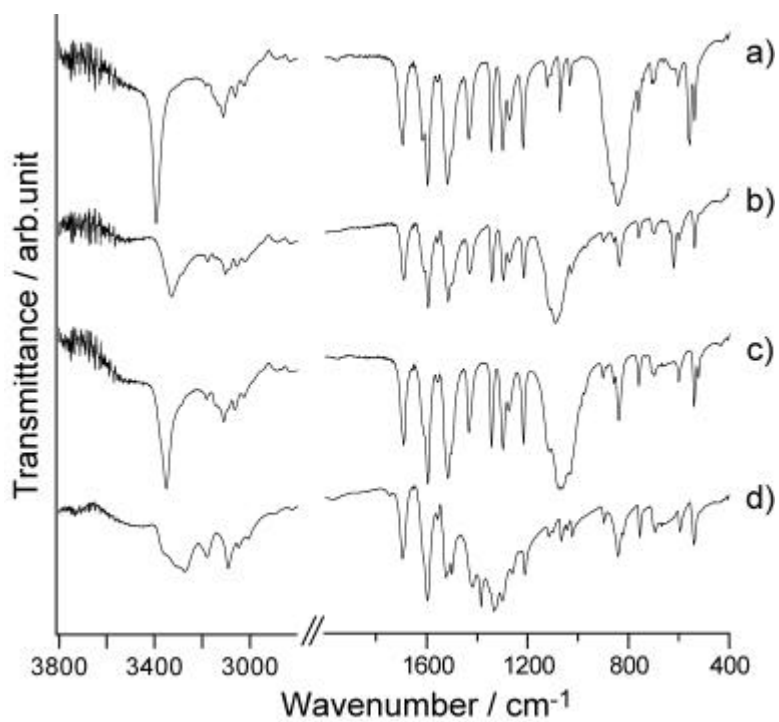


Figure S3. IR spectra of (a)  $\{[\text{Ag}(4\text{-pia})](\text{PF}_6)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{PF}_6^-$ ), (b)  $\{[\text{Ag}(4\text{-pia})](\text{ClO}_4)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{ClO}_4^-$ ), (c)  $\{[\text{Ag}(4\text{-pia})](\text{BF}_4)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{BF}_4^-$ ) and (d)  $\{[\text{Ag}(4\text{-pia})](\text{NO}_3)\}_n$  ( $\mathbf{1} \dot{\text{E}} \text{NO}_3^-$ ) at room temperature.

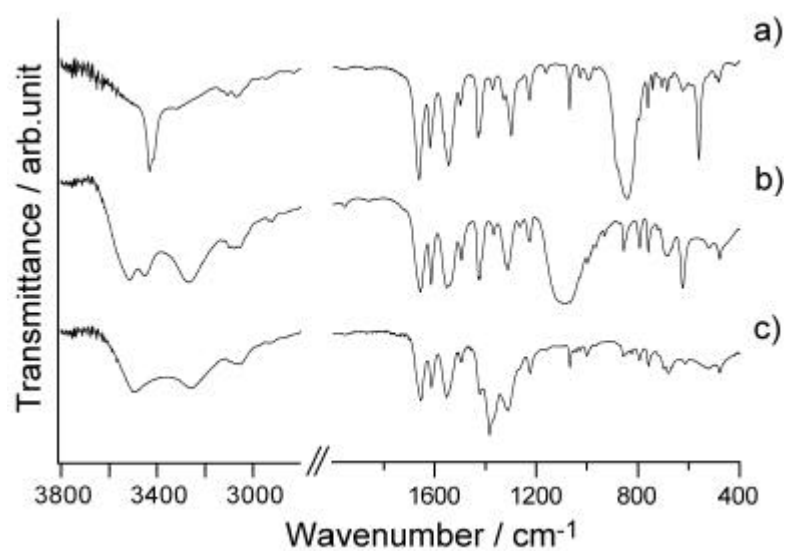


Figure S4. IR spectra of (a)  $\{[\text{Ag}(4\text{-pmia})](\text{PF}_6)\}_n$  ( $2 \hat{=} \text{PF}_6^-$ ), (b)  $\{[\text{Ag}(4\text{-pmia})](\text{ClO}_4)\cdot\text{H}_2\text{O}\}_n$  ( $2 \hat{=} \text{ClO}_4^- \cdot \text{H}_2\text{O}$ ) and (c)  $\{[\text{Ag}(4\text{-pmia})](\text{NO}_3)\cdot\text{H}_2\text{O}\}_n$  ( $2 \hat{=} \text{NO}_3^- \cdot \text{H}_2\text{O}$ ) at room temperature.

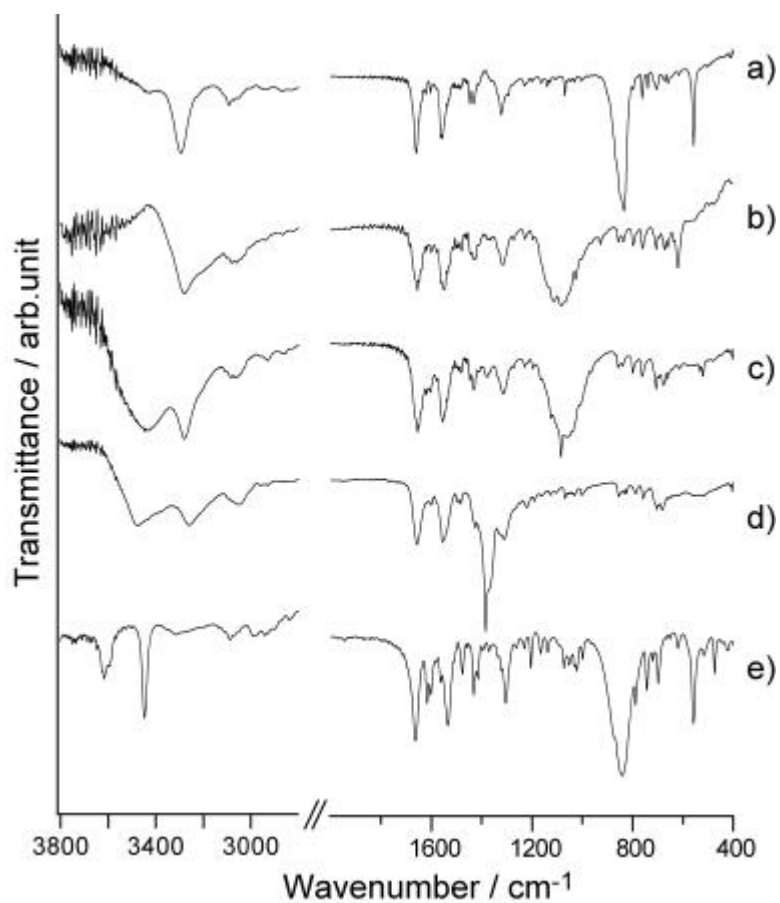


Figure S5. IR spectra of (a)  $\{[\text{Ag}(3\text{-pmia})](\text{PF}_6)\}_n$  (**3**  $\dot{\text{E}}$   $\text{PF}_6^-$ ), (b)  $\{[\text{Ag}(3\text{-pmia})](\text{ClO}_4)\}_n$  (**3**  $\dot{\text{E}}$   $\text{ClO}_4^-$ ), (c)  $\{[\text{Ag}(3\text{-pmia})](\text{BF}_4)\}_n$  (**3**  $\dot{\text{E}}$   $\text{BF}_4^-$ ), (d)  $\{[\text{Ag}(3\text{-pmia})](\text{NO}_3)\cdot\text{H}_2\text{O}\}_n$  (**3**  $\dot{\text{E}}$   $\text{NO}_3^- \cdot \text{H}_2\text{O}$ )

and (e)  $\{[\text{Ag}(4\text{-pmna})](\text{PF}_6)\cdot\text{MeOH}\}_n$  (**1**  $\dot{\text{E}}$   $\text{PF}_6^- \cdot \text{MeOH}$ ) at room temperature.

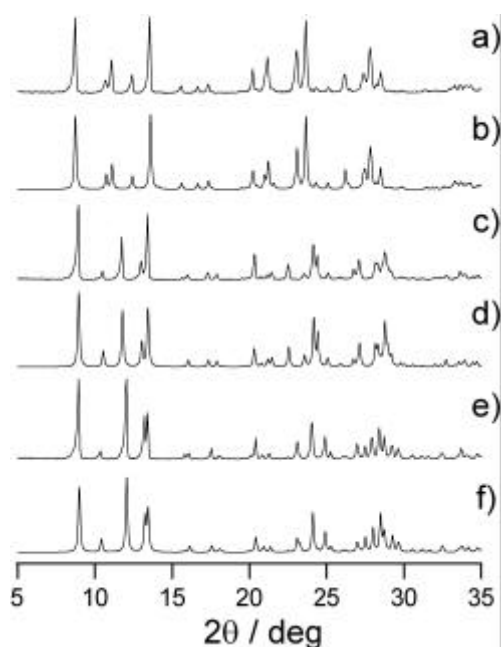


Figure S6. XRPD patterns of (a) as prepared  $\{[\text{Ag}(4\text{-pia})](\text{PF}_6)\}_n$  (**1**  $\dot{\text{E}}$   $\text{PF}_6^-$ ), (b) simulation of **1**  $\dot{\text{E}}$   $\text{PF}_6^-$ , (c) as prepared  $\{[\text{Ag}(4\text{-pia})](\text{ClO}_4)\}_n$  (**1**  $\dot{\text{E}}$   $\text{ClO}_4^-$ ), (d) simulation of **1**  $\dot{\text{E}}$   $\text{ClO}_4^-$ , (e) as prepared  $\{[\text{Ag}(4\text{-pia})](\text{BF}_4)\}_n$  (**1**  $\dot{\text{E}}$   $\text{BF}_4^-$ ) and (f) simulation of **1**  $\dot{\text{E}}$   $\text{BF}_4^-$ . Since it was not succeeded in obtaining the microcrystalline sample **1**  $\dot{\text{E}}$   $\text{NO}_3^-$ , single crystals were used for some experiments. The simulation patterns were based on the single X-ray analyses.



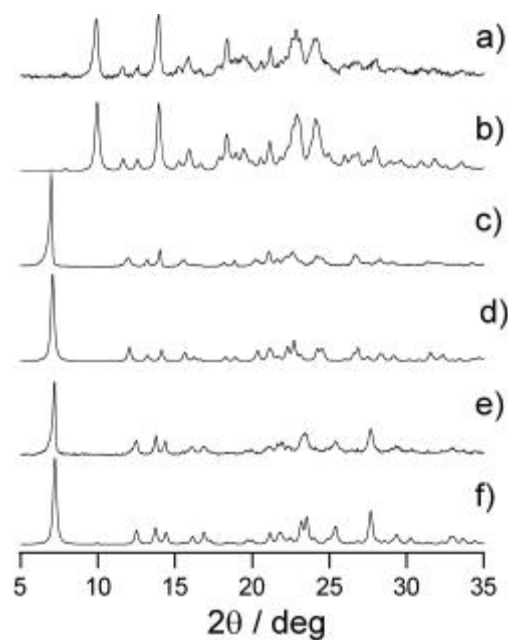


Figure S7. XRPD patterns of (a) as prepared  $\{[\text{Ag}(4\text{-pmia})](\text{PF}_6)\}_n$  (**2**  $\dot{\text{E}}$   $\text{PF}_6^-$ ), (b) simulation of **2**  $\dot{\text{E}}$   $\text{PF}_6^-$ , (c) as prepared  $\{[\text{Ag}(4\text{-pmia})](\text{ClO}_4)\cdot\text{H}_2\text{O}\}_n$  (**2**  $\dot{\text{E}}$   $\text{ClO}_4^- \cdot \text{H}_2\text{O}$ ), (d) simulation of **2**  $\dot{\text{E}}$   $\text{ClO}_4^- \cdot \text{H}_2\text{O}$ , (e) as prepared  $\{[\text{Ag}(4\text{-pmia})](\text{NO}_3)\cdot\text{H}_2\text{O}\}_n$  (**2**  $\dot{\text{E}}$   $\text{NO}_3^- \cdot \text{H}_2\text{O}$ ) and (f) simulation of **2**  $\dot{\text{E}}$   $\text{NO}_3^- \cdot \text{H}_2\text{O}$ . The simulation patterns were based on the single X-ray analyses.

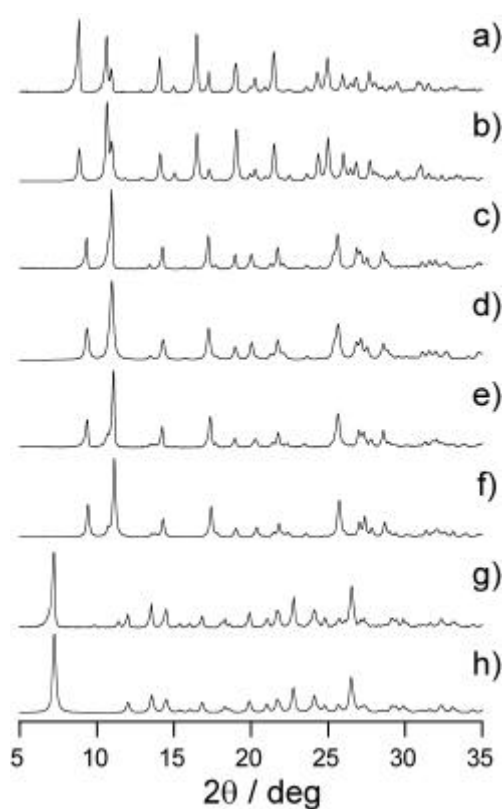


Figure S8. XRPD patterns of (a) as prepared  $\{[\text{Ag}(3\text{-pmia})](\text{PF}_6)\}_n$  (**3**  $\hat{=}$   $\text{PF}_6^-$ ), (b) simulation of **3**  $\hat{=}$   $\text{PF}_6^-$ , (c) as prepared  $\{[\text{Ag}(3\text{-pmia})](\text{ClO}_4)\}_n$  (**3**  $\hat{=}$   $\text{ClO}_4^-$ ), (d) simulation of **3**  $\hat{=}$   $\text{ClO}_4^-$ , (e) as prepared  $\{[\text{Ag}(3\text{-pmia})](\text{BF}_4)\}_n$  (**3**  $\hat{=}$   $\text{BF}_4^-$ ), (f) simulation of **3**  $\hat{=}$   $\text{BF}_4^-$ , (g) as prepared  $\{[\text{Ag}(3\text{-pmia})](\text{NO}_3)\cdot\text{H}_2\text{O}\}_n$  (**3**  $\hat{=}$   $\text{NO}_3^- \cdot \text{H}_2\text{O}$ ) and (h) simulation of **3**  $\hat{=}$   $\text{NO}_3^- \cdot \text{H}_2\text{O}$ . The simulation patterns were based on the single X-ray analyses.

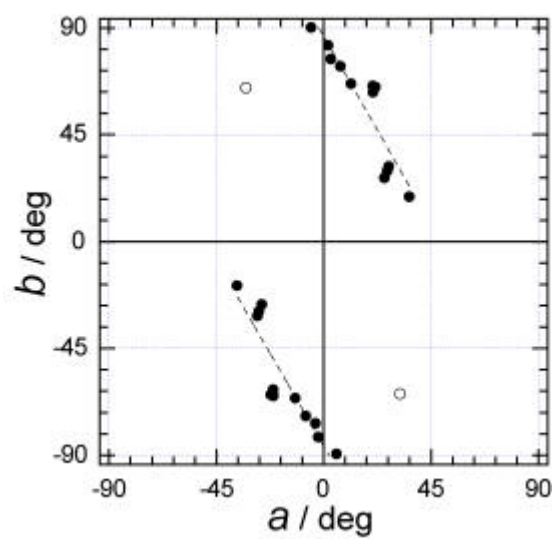


Figure S9. The plots of  $b$  versus  $a$ . Open circles are corresponding to those in  $\{[\text{Ag}(4\text{-pmia})](\text{PF}_6)_n\}_n$  ( $2 \leq n \leq 6$ ). Other angles (filled circles) appear a tendency for a large value of  $a$  to be accompanied by a small value of  $b$ .

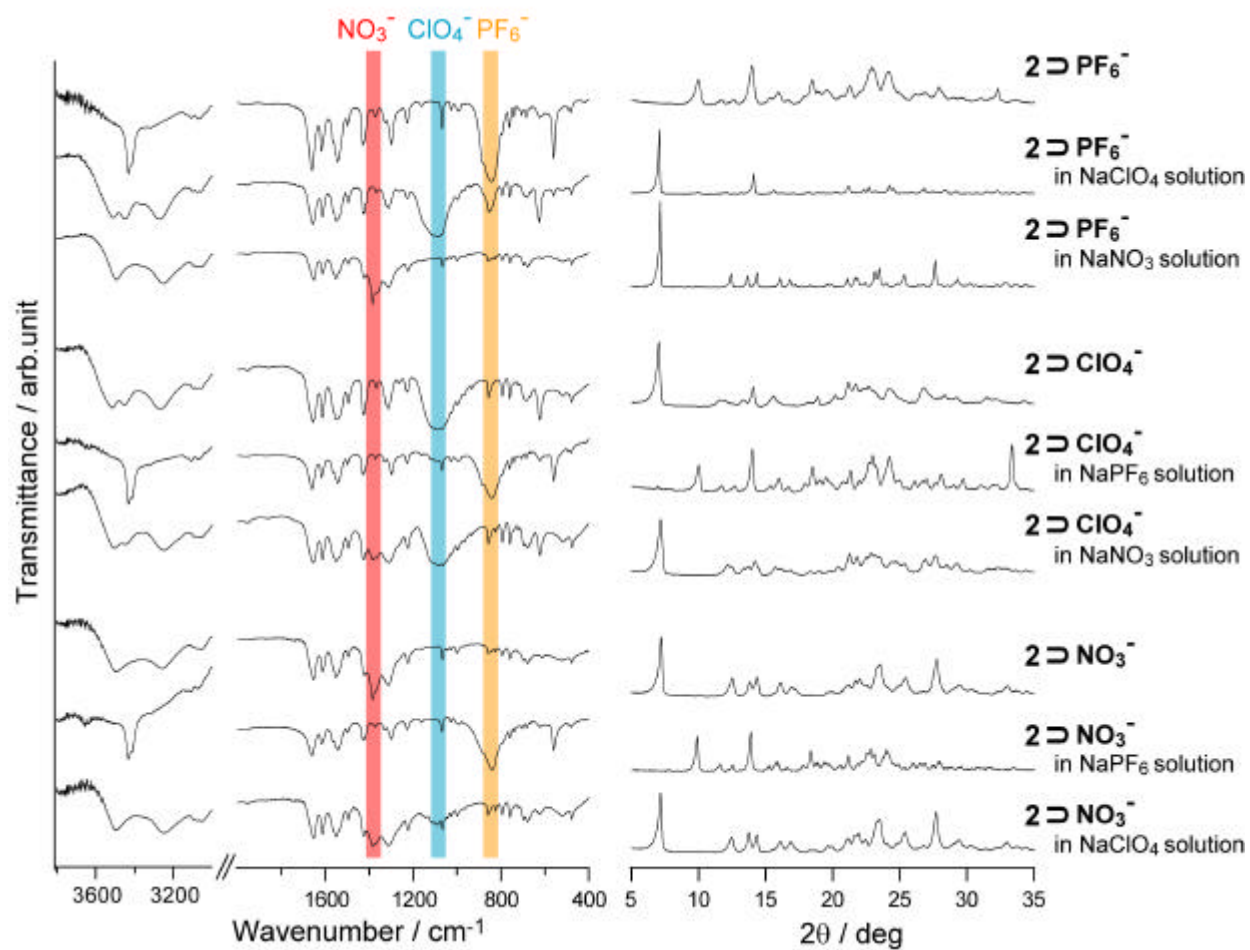
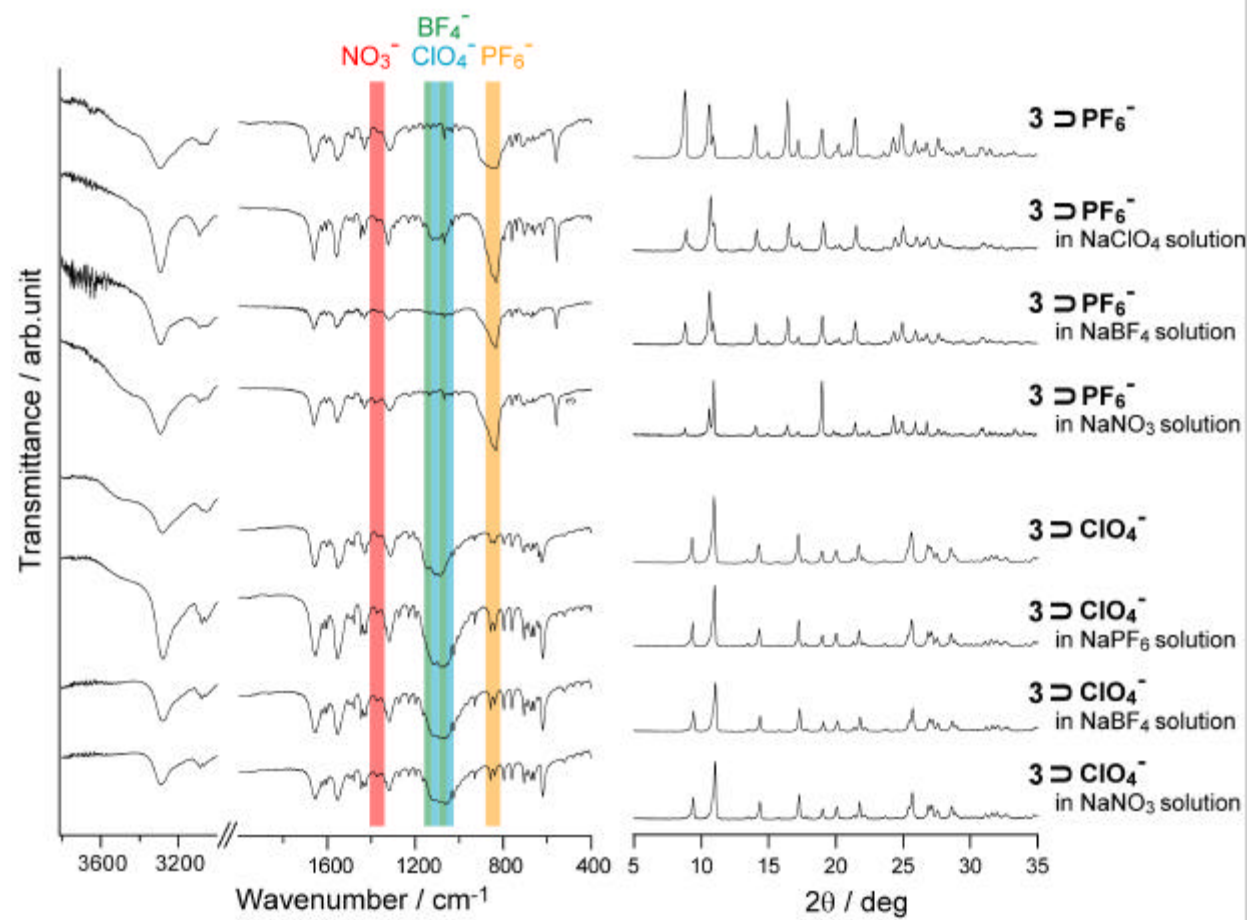


Figure S10. XRPD patterns (left) and IR spectra (right) at room temperature for anion exchange of compounds **2**. See the text for the detailed information.



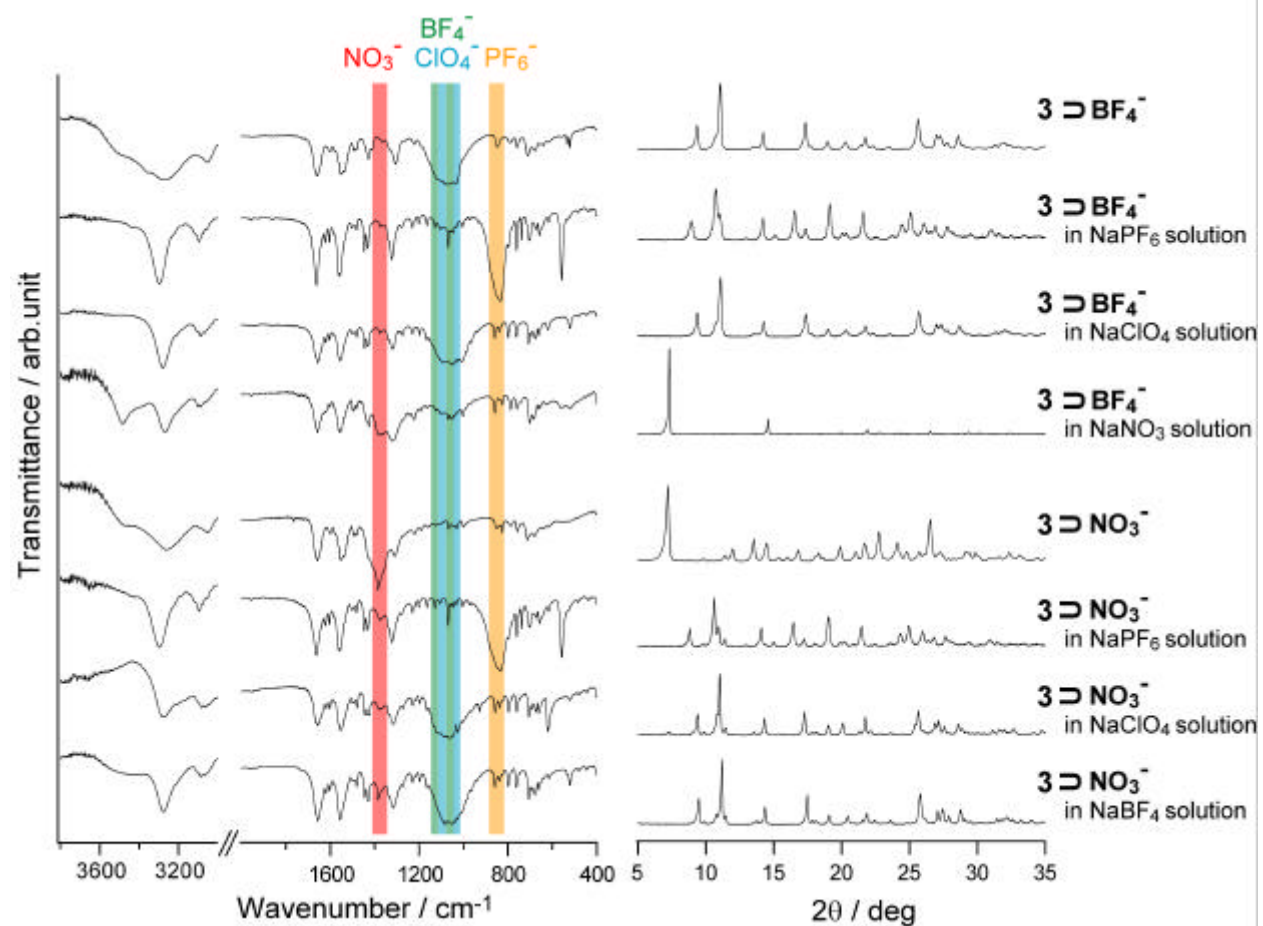


Figure S11. XRPD patterns (left) and IR spectra (right) at room temperature for anion exchange of compounds **3**. See the text for the detailed information.